

Code No.: ME305PC

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**CMR ENGINEERING COLLEGE: : HYDERABAD**  
**UGC AUTONOMOUS**  
**II-B.TECH-I-Semester End Examinations (Supply) - February - 2023**  
**THERMODYNAMICS**  
**(MECH)**

[Time: 3 Hours]

[Max. Marks: 70]

**Note:** This question paper contains two parts A and B.

Part A is compulsory which carries 20 marks. Answer all questions in Part A.

Part B consists of 5 Units. Answer any one full question from each unit. Each question carries 10 marks and may have a, b, c as sub questions.

**PART-A**

**(20 Marks)**

1. a) Define irreversible process. Give examples. [2M]
- b) State zeroth law of thermodynamics. [2M]
- c) Write the differential form of the steady flow energy equation. [2M]
- d) State Kelvin-Planck statement. [2M]
- e) Define p-v-T surface and state its significance. [2M]
- f) Write down the equation of state of an ideal gas. [2M]
- g) Define Specific humidity and relative humidity. [2M]
- h) State Avagadro's law. [2M]
- i) Differentiate between Sterling and Ericsson Cycles. [2M]
- j) State elements of vapour compression refrigeration system. [2M]

**PART-B**

**(50 Marks)**

2. A fluid at a pressure of 3 bar and with specific volume of  $0.18 \text{ m}^3/\text{kg}$ , contained in a cylinder behind a piston expands reversibly to a pressure of 0.6 bar according to a law,  $p = (C/v^2)$  where C is a constant. Calculate the work done by the fluid on the piston. [10M]

**OR**

- 3.a) Compare Macroscopic and Microscopic approaches in thermodynamic studies. [5M]
- b) Explain about point function and path function with examples. [5M]
- 4.a) A cyclic heat engine operates between a source temperature of  $800^\circ\text{C}$  and a sink temperature of  $30^\circ\text{C}$  and heat supply to heat engine is 50kW. What is the least rate of the heat rejection per kW net output of the engine? [6M]
- b) Apply first law to a process and a cycle. [4M]

**OR**

- 5.a) Explain in detail about Clausius inequality. [5M]
- b) 3 kg of air is expanded at constant pressure to three times its initial volume. Calculate the change in entropy if the initial temperature of the air is  $300^\circ\text{C}$ . [5M]
6. Steam at a pressure of 5 bar passes into a tank containing water where it gets condensed. The mass and temperature in the tank before the admission of steam are 50 kg and  $20^\circ\text{C}$  respectively. Calculate the dryness fraction of steam as it enters the tank if 3 kg of steam gets condensed and resulting temperature of the mixture becomes  $40^\circ\text{C}$ . Take water equivalent of tank as 1.5 kg. [10M]

**OR**

- 7.a) Explain Throttling process and Free expansion process. [5M]

- b)  $0.3 \text{ m}^3$  of air at pressure 8 bar expands to  $1.5 \text{ m}^3$ . The final pressure is 1.3 bar. [5M]  
Assuming the expansion to be polytropic. Calculate the heat supplied and change of internal energy. Assume  $\gamma=1.4$
- 8.a) A gas mixture consists of 0.4 kg CO, 1.1 kg of  $\text{CO}_2$  and 1.5 kg of  $\text{N}_2$ . Determine [6M]  
i. Mass fraction of each component.  
ii. Mole fraction of each component.  
iii. Gas constant of the mixture.
- b) Write a short note on the Gravimetric Analysis. [4M]
- OR**
- 9.a) State Daltons law of partial pressure. [2M]  
b) Define DBT, WBT, Dew point temperature and degree of saturation. [8M]
10. Derive an expression for efficiency of Otto cycle. Draw p-V and T-s diagrams. [10M]
- OR**
11. Explain the working of Bell-Coleman cycle and derive the expression for COP. [10M]

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